

TRICKS of the NATURE

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Ten short stories about accidental equalities or cancellations
which lead to surprises as well as to useful applications.

Inspired by

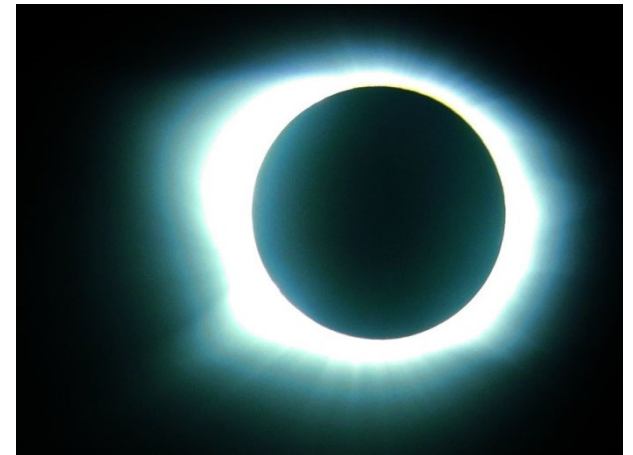
- # R.Peierls, *Surprises in Theoretical Physics*, Princeton University Press 1979.
- # R.Peierls, *More Surprises in Theoretical Physics*, Princeton University Press 1991.
- # V.F.Weisskopf, *Search for Simplicity*, Am.J.Phys **53** (1985) 19,109,206,304,399,522,618,814,940,1140; **54** (1986)13,110.
- # Jo Hermans, *Physics in Daily Life*, EDP Sciences 2012.

1. TOTAL SOLAR ECLIPSE

The apparent sizes of SUN and MOON are more or less the same. Therefore we see occasionally a total solar eclipse or an anular eclipse (when the moon is far or the Sun is near).

$$\alpha_{\text{sun}} = 2r_{\text{sun}}/R_{\text{sun}} = 1.4 \times 10^9 \text{m} / 150 \times 10^9 \text{m} \approx 0.01 \approx 30'$$

$$\alpha_{\text{moon}} = 2r_{\text{moon}}/R_{\text{moon}} = 3.5 \times 10^6 \text{m} / 380 \times 10^6 \text{m} \approx 0.01 \approx 30'$$



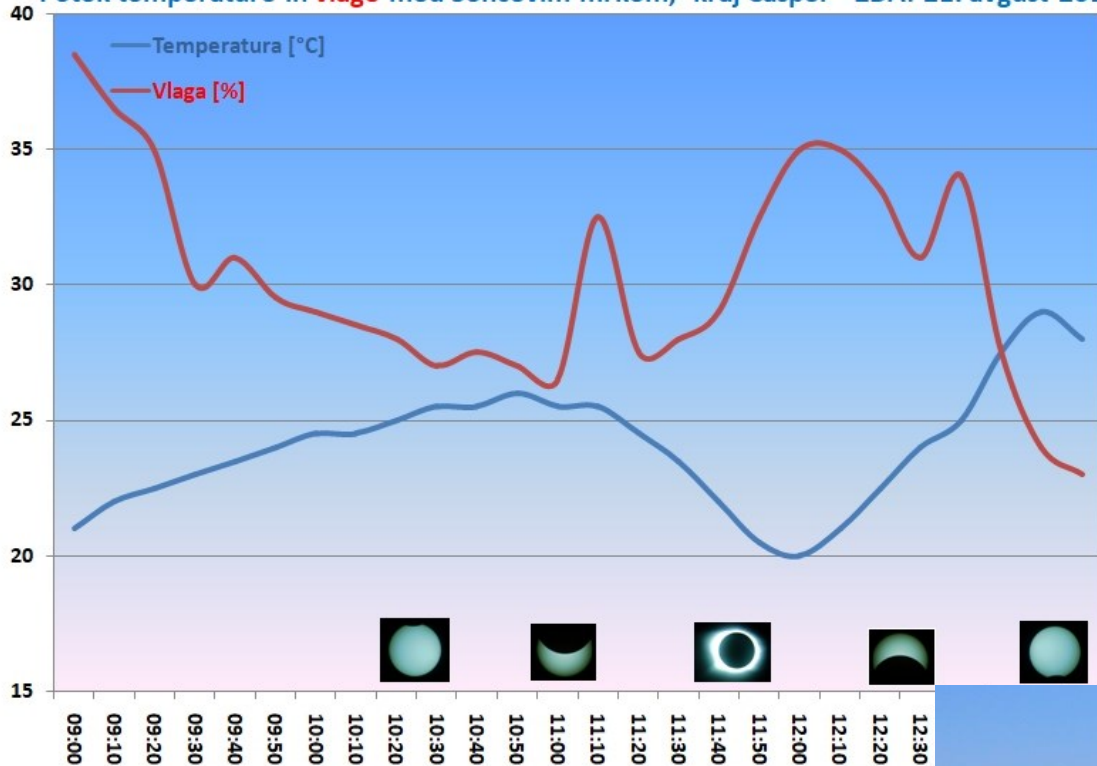
Casper, Wyoming, USA,
August 21, 2017

10:22 – 11:42:39

– 11:45:09 – 13:09

(photo: Zorko Vičar)

Potek temperature in vlage med Sončevim mrkom, kraj Casper - ZDA: 21. avgust 2017



Our enthusiastic „Solar expedition“ 2017 counted 16 participants.

The temperature dropped by 6°, then it rose by 9° (Zorko Vičar)



We visited several national parks in the Western USA



Rocky Mountains NP
Kolorado

Yellowstone NP
Wyoming/
Montana / Idaho

Grand Teton NP
Wyoming

Great Salt Lake
Utah



Zion NP
Utah



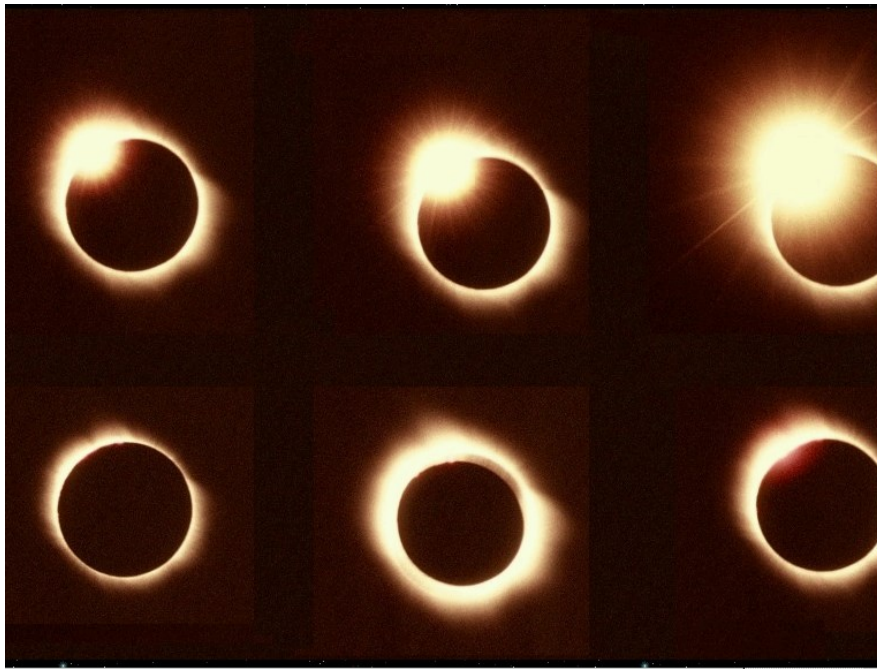
Bryce Canyon NP
Utah



Horseshue Bend
(Colorado river)
Arizona



Grand Canyon
Arizona



TOTAL SOLAR
ECLIPSE IN
TURKEY
29.3.2006

PHOTO: Zorko Vičar
and his team



Matjaž je ugotovil, da so bile nekatere rožice, cvetovi, pred in po mrku odprte



Med mrkom pa so bile nekatere rožice (cvetovi) zaprte, v centralnem delu Evrope, lahko spremljali obnašanje živali (črčki, kokoši, race, psi itn)

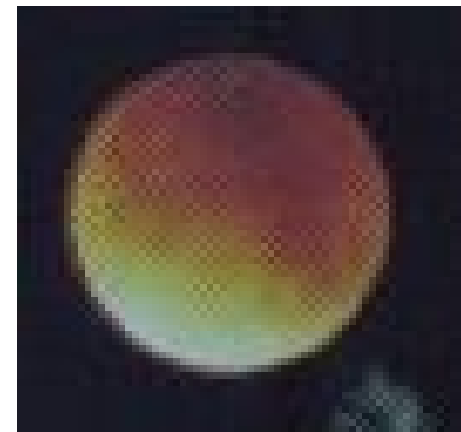


Vladimir Brezar: Solar eclipse in Antalya, Turkey 2006



Ivan Generalić: Pomrčenje sunca (1961)

2. TOTAL LUNAR ECLIPSE 21.1.2000



The brilliance of the moon
Was estimated by comparison
with the church

$$j_{in} = 10 \text{ lx}$$

(1000 cd reflector at 10 m).

Albedo:

$$a(\text{wall}) = 0.6, \quad a(\text{roof}) = 0.04$$

$$j_{out}(\text{wall}) = 6 \text{ lx}$$

$$j_{out}(\text{roof}) = 0.4 \text{ lx}$$

$$j_{out}(\text{moon}) = \text{inbetween}$$

Surprise: Why is the eclipsed Moon so brilliant?

Answer: The air layer around the illuminated ring facing the Moon acts as a ring lense and the Moon happens to be near the focus.

For a spherical lense $f = R/2(n-1)$. With $R(\text{earth}) = 6400 \text{ km}$ and $n = 1,0003$ we get $f = 10^7 \text{ km}$. TOO MUCH!

For a ring lense $f = \sqrt{Rh/2\pi}/(n-1)$.

$h = 7.7 \text{ km}$ is the effective thickness of Earth atmosphere (exponential decrease assumed). We get $f = 300\,000 \text{ km}$. FINE!

Illumination of the Moon

Sun illuminates Earth with $j_0 = 100\,000 \text{ lx}$.

The incoming light on the ring is distributed around the focus to an area of Earth cross section or more (an estimate from drawing the rays).

The Moon gets

$$j_{\text{in}} < j_0 \times (2\pi Rh / \pi R^2) = j_0 (2h/R) = 240 \text{ lx}.$$

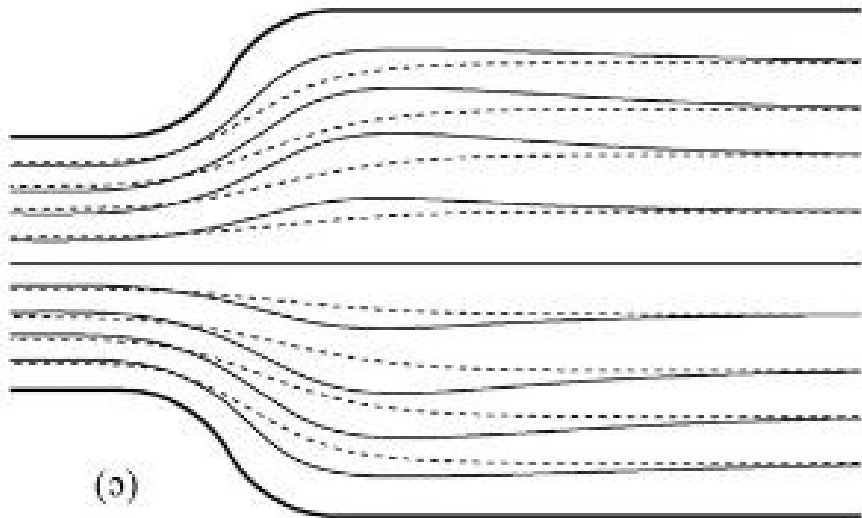
The Moon shines (albedo = 0,08)

$J_{\text{out}} < 20 \text{ lx}$ (or less, depending on the weather).

3. TROUT IN A DIVERGENT CURRENT

I was surprised to see a trout swimming at a fixed point in a rapid river. **The explanation is easy:**

The trout was swimming at a constant speed, the same but opposite as the water. If she came forward, she was carried back by the faster water. If she stayed backward, she overtook the slower water. Note that we are also used to walk or swim at a constant speed.



4. ENERGY SOURCE OXYGEN

The oxygen double bond (O=O) is a covalent bond. The bond energy is only as large as a single bond energy for light atoms in the (2s, 2p) shells and in the hydrogen molecule.

H ₂	bonds=1	$\Delta E/\text{bond} = 106 \text{ kcal/mol} = 4.6 \text{ eV}$
-C-C-	bonds=1	$\Delta E/\text{bond} = 100 \text{ kcal/mol} = 4.3 \text{ eV}$
CH ₄	bonds=4	$\Delta E/\text{bond} = 98 \text{ kcal/mol} = 4.3 \text{ eV}$
H ₂ O	bonds=2	$\Delta E/\text{bond} = 96 \text{ kcal/mol} = 4.2 \text{ eV}$
CO ₂	bonds=4	$\Delta E/\text{bond} = 106 \text{ kcal/mol} = 4.6 \text{ eV}$
O ₂	bonds=2	$\Delta E/\text{bond} = 59 \text{ kcal/mol} = 2.4 \text{ eV}$

One speaks of fossil fuel energy sources and means coal, gas and oil. The energy is, however, stored in atmospheric oxygen!

Example: $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$. The number of covalent bonds (8) remains constant, as the four weak oxygen bonds are replaced by four stronger ones. $\Delta E = 190 \text{ kcal} \approx 4 \times 2 \text{ eV}$.

Photosynthesis, which separates oxygen from carbon, has stored the energy in the weak bond in oxygen.

5. DIRECTIONALITY OF THE COVALENT BOND IN p SHELL

H₂O 104.5°

NH₃ 106.7°

CH₄ 109.5°

H₂S 92.2°

PH₃ 93.3°

SiH₄ 109.5°

H₂Se 91.0°

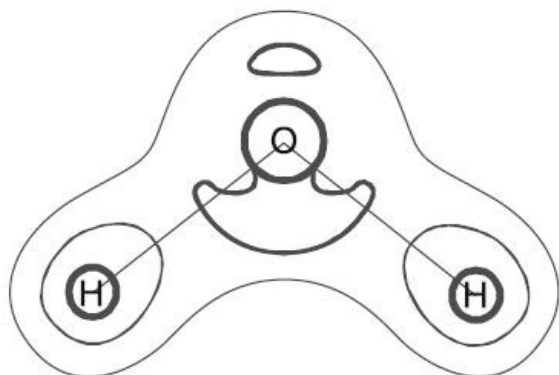
AsH₃ 91.8°

GeH₄ 109.5°

H₂Te 89.5°

SbH₃ 91.3°

SnH₄ 109.5°



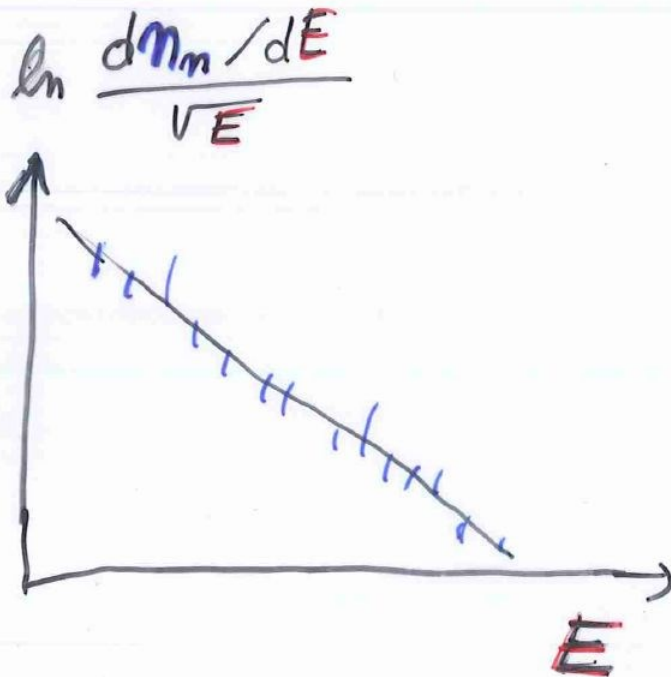
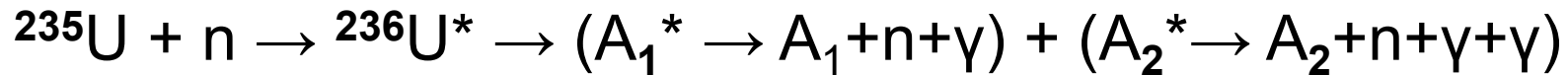
Water molecule.

Two orthogonal 2p orbitals are maximally correlated at an angle of 90°. A superposition of 2s-2p may be at any angle between 90° and 120°. Admixture of 2s is less favoured by electrons but more by repulsion of ions.

1. Large electric dipole moment ($\mu_e = 0.068 e a_0$) → High refraction
2. Large electric dipole moment → Good solvent
3. Forms rings → Ice is lighter than water
4. Forms rings → Large specific heat (9k instead of $6 \times k/2 = 3k$)

6. EVAPORATION OF NEUTRONS FROM FISSION PRODUCTS

During the fission of uranium sufficient number of neutrons for the chain reaction is evaporated. The temperature of the two fission fragments is just high enough to evaporate 2 - 3 neutrons.



Slope = „thermometer“

$$dn_n/dE = \sqrt{E} \exp(-E/kT)$$

$$kT = 1.3 \text{ MeV}$$

Richardson: $n_n = (m_n/2\pi^2\hbar^3)(kT)^2 \exp(-W_i/kT) \times S \times t$

$W_i = 7.5 \text{ MeV},$

$S =$ surface area of both nuclei, $t =$ lifetime

Stefan-Boltzmann: $E_\gamma = \sigma T^4 \times S \times t$

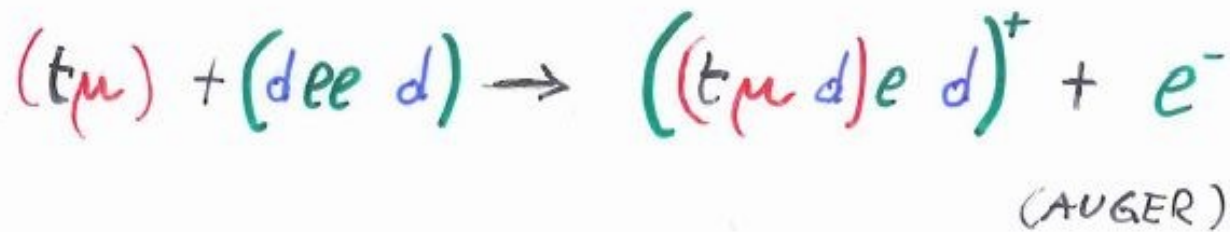
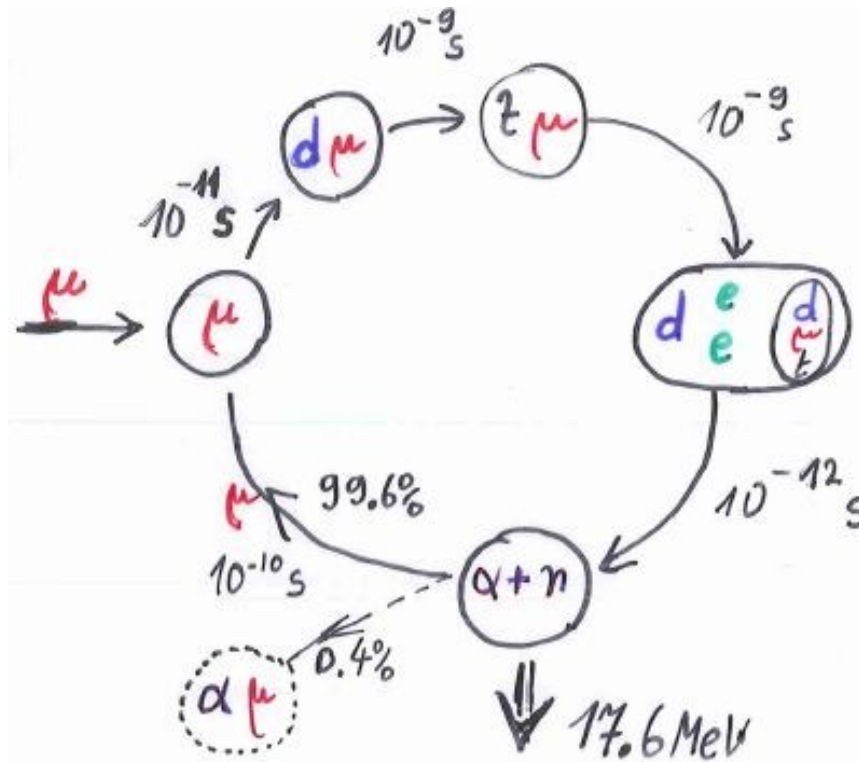
$(n_n \times 3kT/2) / E_\gamma \approx 1$

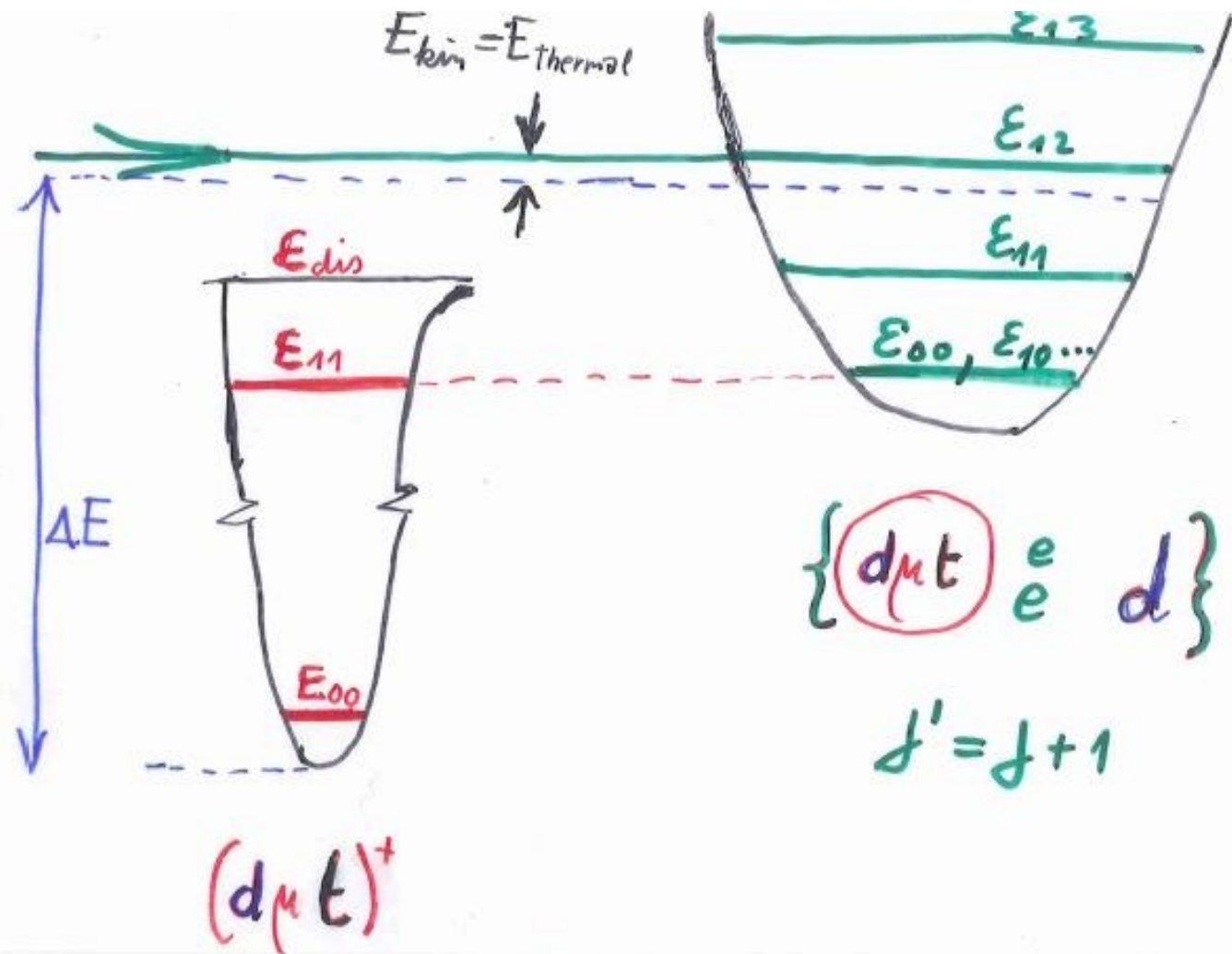
Experiment: $E_n (= n_n \times 3kT/2) = 5 \text{ MeV}, [n_n = 2,6]$
 $E_\gamma = 6 \text{ MeV}$

(These values are consistent with the above formulas and nuclear model assumptions.)

7. MUON CATALYSED FUSION

Muons can catalyse over 150 fusions of deuteron and triton into helium, because the energy balance for the resonant formation of a combined muonic-electronic molecule can be adjusted to meV precision at 0,3 keV.





$$E_{t\mu} + E_{d\mu e d} + E_{kin} = E_{(d\mu t)ee d} + E_{jV} + E_{j'V}$$

$$\Delta E + E_{kin} = E_{jV} + E_{j'V}$$

8. THE SYNTHESIS OF CARBON FROM HELIUM IS HELPED BY RESONANCES

The synthesis of carbon and heavier elements in older stars is possible because an excited state in ^{12}C is just at the right place to enhance the resonant synthesis.

„Without this resonance we would not be here!“ (Fred Hoyle)

Middle-aged stars (our Sun): $4\ ^1\text{H} \rightarrow\ ^4\text{He} + 2\ \text{e}^+ + 2\nu$

Old stars ($\rho=10^9\ \text{kg/m}^3$, $T= 10^8\ \text{K}$):



$$\rho = 10^9 \text{ kg/m}^3, T = 10^8 \text{ K}, kT = 8.6 \text{ keV}, n_4 = 1.5 \times 10^{35} \text{ m}^{-3}$$

$$\mu_4 + \mu_4 = \mu_8 + \Delta E_8 ; \quad n_8 / n_4 = 6.6 \times 10^{-9}$$

$$\mu_8 + \mu_4 = \mu_{12}^* + \Delta E_{12} ; \quad n_{12}^* / n_4 = 3.7 \times 10^{-27}$$

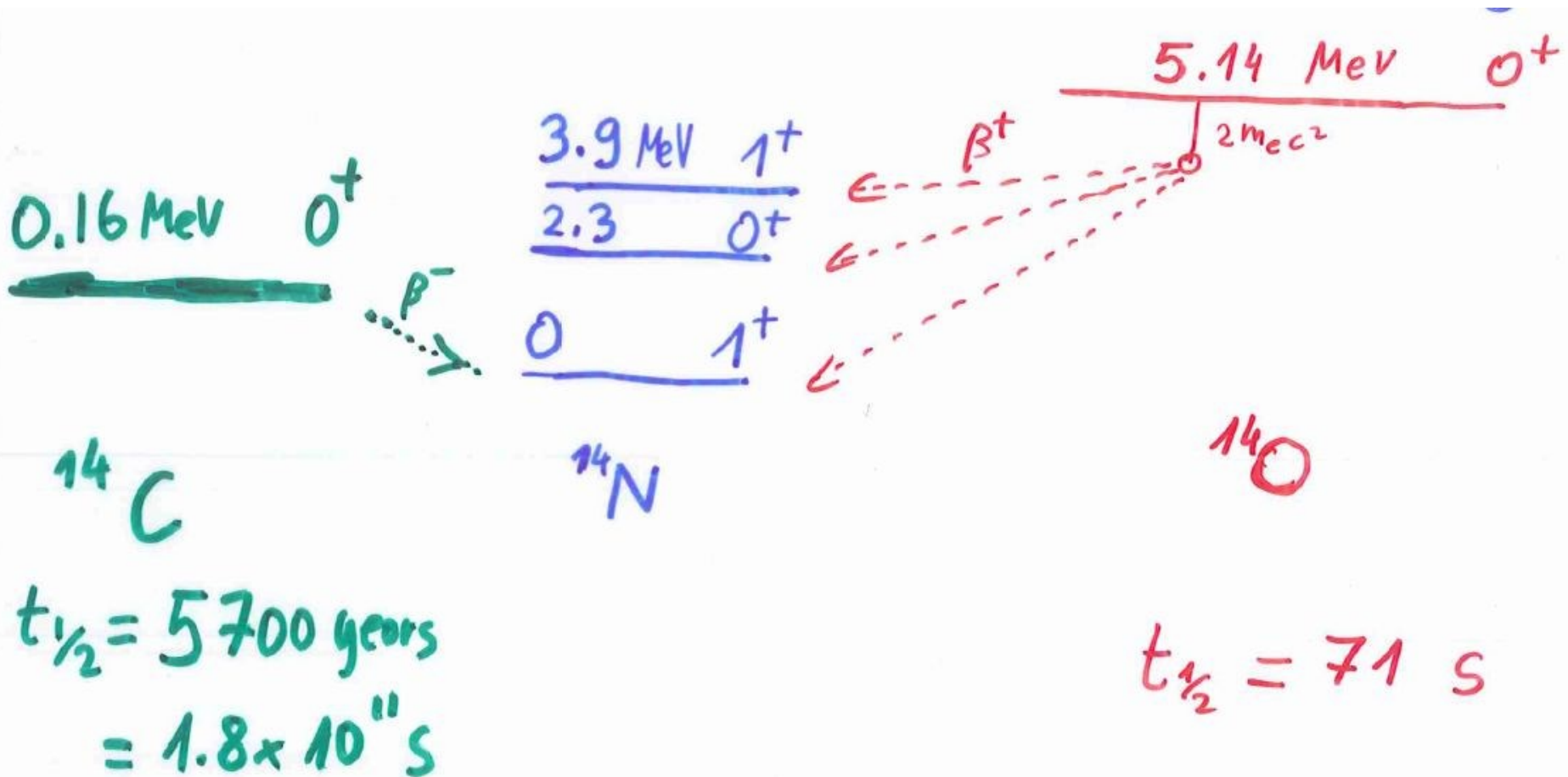
$$d n_{12} / d t = n_{12}^* w, \quad w = 5.6 \times 10^{12} \text{ s}^{-1}$$

$$(d n_{12} / d t) / n_4 = 2.1 \times 10^{-14} \text{ s}^{-1} = (1.5 \times 10^6 \text{ years})^{-1}$$

9. THE ^{14}C NUCLEUS HAS A NINE ORDERS OF MAGNITUDE LONGER HALFLIFE THAN THE MIRROR NUCLEUS ^{14}O



PLEASURE FOR ARCHEOLOGISTS!



The ^{14}C and ^{14}O nuclei can decay into the ground state of ^{14}N through several intermediate states. The accidental destructive interference is almost perfect in ^{14}C , but not in ^{14}O .

Moreover, the ^{14}O decay has a larger phase space, and can also decay into excited states of ^{14}N .

The beta-decay matrix element in $^{14}\text{C} \rightarrow ^{14}\text{N}$ is retarded to 0.1% while in $^{14}\text{O} \rightarrow ^{14}\text{N}$ it is retarded to 1%.

It misses the „perfect cancellation“ by 1%.

Anyway, 1% is the precision of isospin symmetry

10. THE MASS OF THE STRANGE QUARK IS SUFFICIENT TO PREVENT THE COLLAPSE INTO STRANGE MATTER

The Pauli Principle would favour a nuclear matter composed of several flavours of quarks (for example **s** quarks in addition to **u** and **d** quarks) if **s** quarks were not sufficiently heavy.

Experimental hint:

There are no $\Lambda\Lambda = uds + uds$ bound states;

The $\Lambda\Lambda^6\text{He}$ decays weakly.

Nuclear matter estimate: $g = 2 \times 2 \times 3 = 12$ (spins, isospins, colours)

$$W = (3/5) (p_F^2/2m), N = 3n_q$$

$$E/N = 3m + 3W = 3(223 + 87) \text{ MeV} = 930 \text{ MeV}$$

Strange matter estimate: $g = 2 \times 3 \times 3 = 18$ (spins, flavours, colours)

$$m_s - m \approx m_\Lambda - m_N \approx m_\Xi - m_\Lambda \approx 180 \text{ MeV (from experiment)}$$

$$E/N = [2(223 + 87 \sqrt[3]{12/18}) + 1(403 + 87 \sqrt[3]{12/18} (223/403))] \text{ MeV}$$

$$= 1043 \text{ MeV. WE ARE SAFE !!!}$$